



# Abstract #796

English

## Mars Propellant Production Using Ionic Liquids

The use of Ionic Liquids for CO<sub>2</sub> capture and electrolysis has several advantages that make it well-suited for in situ propellant production. These advantages can be exploited for future Mars Sample Return and human exploration missions. Ionic liquids are capable of capturing a large mole fraction of CO<sub>2</sub> and can serve as an electrolyte for direct electrolysis of CO<sub>2</sub> and H<sub>2</sub>O to CH<sub>4</sub> and O<sub>2</sub>. We expect this method to be more energy efficient (~25% less energy), require fewer processing steps, and ~50% less mass, as compared to CO<sub>2</sub> freezing/methanation/water electrolysis. We are verifying this process which would greatly reduce power, mass, and complexity through the use of a single vessel for CO<sub>2</sub> capture and electrolysis to propellant. Electrolysis of CO<sub>2</sub> + H<sub>2</sub>O in ionic liquids to CH<sub>4</sub> and O<sub>2</sub> has not been demonstrated before, although electrolysis of the gases to methane and oxygen has been performed with a copper catalyst and Nafion membranes. This presentation will report the results of a joint Kennedy Space Center (KSC), Marshall Spaceflight Center (MSFC), and Mercer University project to identify and test ILs with the correct properties (CO<sub>2</sub> sorption capacity, low viscosity, high conductivity, and electrochemical window) and electrocatalysts.

French

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